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Title:

ANTI-INFLAMMATORY 4,5-DIARYL- ALPHA -(POLYFLUOROALKYL)-1H-PYRROLE-2-METHANAMINES

Abstracted Patent

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ABSTRACT:

4,5-Diaryl- alpha -(polyfluoroalkyl)-1H-pyrrole-2-methanamines such as 4,5-bis(4-fluorophenyl)-alpha alpha -bis-(trifluoromethyl)-1H-pyrrole-2-methanamine, useful in treatment of inflammation.

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- 64) Antiinflammatory 4,5-diaryl-alpha-(polyfluoroalkyl)-1H-pyrrole-2-methanamines.
- (i) 4,5-Diaryl- α -(polyfluoroalkyl)-1H-pyrrole-2-methanamines such as 4,5-bis(4-fluorophenyl)- α , α -bis-(tri-fluoromethyl)-1H-pyrrole-2-methanamine, useful in treatment of inflammation.

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Title

BP-6181-A

Antiinflammatory 4,5-Diaryl-C-(polyfluoroalkyl)-1H-pyrrole-2-methanamines

Background of the Invention

This invention relates to antiinflammatory pyrroles.

J. Szmuszko vicz et al., J. Med. Chem., 9, 527 (1966) describe synthesis and biological activity of a clinically tested antiinflammatory agent of the formula

Yoshida et al., U.S. Patent 3,709,906 disclose 5-alkyl-2,3-diphenylpyrrole derivatives which are useful as antiinflammatory agents.

There is a continuing need for safe and effective antiinflammatory agents. Inflammation is a disease process characterized by redness, fever, swelling, and pain. Arthritis, in its various forms, is the most prevalent, chronic, and severe of the inflammatory diseases. Traumatic injury and in-25 fection also involve inflammation, and antiinflammatory drugs are often used in their treatment. The usefulness of most commercial antiinflammatories is limited because of toxicity and adverse sideeffects. Many produce gastric irritation and other **30** effects, such as changes in blood cells and central nervous system. Adreno-cortical steroids produce gastric irritation and suppression of normal adrenal function.

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The present invention results from efforts to develop new anti-arthritic compounds with good antiinflammatory activity and minimal side effects that could be more effective in treating arthritis then presently available drugs.

In addition to antiinflammatory properties, some compounds of this invention have demonstrated analgesic activity in a test procedure. This additional property is desirable in treatment of arthritis or related diseases; however, such compounds can be employed solely to alleviate pain.

Summary of the Invention

This invention relates to compounds of Formula I, pharmaceutical compositions containing them, and methods of use of these compounds to treat arthritis.

$$\begin{array}{c|c}
R_2 & R_1 \\
R_3 & R_4 \\
R_6 & R_5
\end{array}$$

25 wherein

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where

X = H, F, Cl, Br, C_1-C_2 alkyl, C_1-C_2 alkoxy, $di(C_1-C_2$ alkyl)amino or $CH_3S(0)_n$ where n=0, 1 or 2; and

Y = H, F or Cl; with the proviso that when Y is F or Cl, then X is F or Cl;

 R_4 and R_5 independently = H, CF_3 , CF_2H , CF_2Cl , $CFCl_2$ or CF_2CF_3 , with the proviso that no more than one of R_4 and R_5 can be H; and the further proviso that no more than one of R_4 and R_5 can be CF_2CF_3 ;

R₆ and R₇ independently = H, C₁-C₆ alkyl, benzyl or benzyl substituted with up to two atoms selected from the group consisting of F, Cl, Br, NO₂, and CF₃; with the proviso that when R₄ or R₅ is H, then R₇ must be H also; or a pharmaceutically suitable acid addition salt

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thereof. Preferred Compounds

Preferred compounds for utility considerations or ease of synthesis are those in which, independently:

(a) $R_1 = H \text{ or } CH_3; \text{ or }$

25 (b)
$$R_2$$
 and R_3 independently = X

where X = Br, Cl, F, CH_3^0 or $(CH_3)_2^N$ and Y = H; or

- 30 (c) R_4 and $R_5 = CF_3$; or
 - (d) $R_6 = H \text{ or } CH_3$; or
 - (e) $R_7 = H \text{ or } CH_3$.

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More preferred compounds for the same reasons are those in which:

R₁ = H or CH₃; and R₂ and R₃ independently = X

where X = Br, CI, F, CH_3O or $(CH_3)_2N$ and

Y = H; and

 R_A and $R_5 = CF_3$; and

10 $R_6 = H \text{ or } CH_3$; and $R_7 = H \text{ or } CH_3$.

Specifically preferred for the same reasons are:

- (a) 4,5-Bis(4-fluorophenyl)-d,d-di(trifluoromethyl)lH-pyrrole-2-methanamine;
- 15 (b) 4,5-Bis(4-fluorophenyl)-N-methyl-d,d-di(trifluoro-methyl)-lH-pyrrole-2-methanamine; and
 - (c) 4,5-Bis(4-fluorophenyl)-N,1-dimethyl-QC-di(tri-fluoromethyl)-lH-pyrrole-2-methanamine.

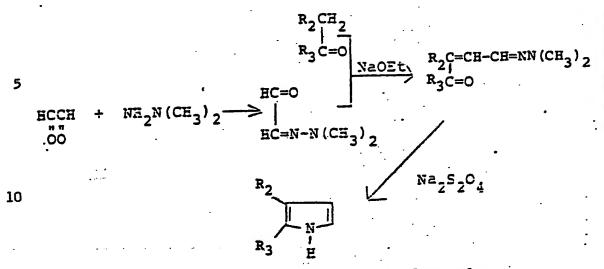
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Synthesis ·

The compounds of this invention can be prepared from 2,3-diarylpyrroles. One method of preparation of 2,3-diarylpyrroles involves reaction of substituted a-aminodeoxybenzoins with acetylene diesters, followed by hydrolysis and decarboxylation according to the procedure used by J. Szmuszkovicz et al., J. Med. Chem., 9, 527 (1966) and by U.S. Patent 3,462,451, the disclosures of which are hereby incorporated by reference, for the synthesis of 2,3-bis(4-methoxyphenyl)pyrrole. (Scheme I).

Another method of preparation of 2,3-diarylpyrroles
is a modification of the procedure of T. Severin and
H. Poehlmann, Chem. Ber., 110, 491 (1977), hereby
incorporated by reference, which describes the preparapreparation of monoaryl pyrroles. By using substituted desoxybenzoins, the desired 2,3-diarylpyrroles
are formed (Scheme II).

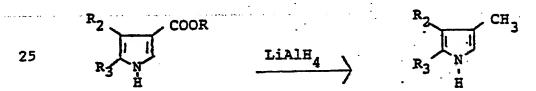


Preparation of 2,3-diaryl-4-alkylpyrroles

can be accomplished by several methods. First,

4,5-diarylpyrrole-3-carboxylate esters, prepared,
for instance, by the method of A. M. van Leusen
et al., Tet. Letters, 5337 (1972) can be reduced
to the 2,3-diaryl-4-methylpyrroles by lithium

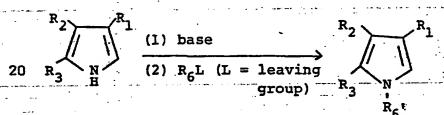
aluminum hydride [following the general procedure
of R. L. Hinman and S. Theodoropulos, J. Org. Chem.,
28, 3052 (1963)], hereby incorporated by reference.



Secondly, 2,3-diaryl-4-alkylpyrroles can be prepared by the general procedure of N. Engel and W. Steglich, Angew. Chem. Int. Ed. Engl., 17, 676 (1978), hereby incorporated by reference, from N-allylcarboxamides.

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1-Alkyl-2,3-diarylpyrroles can be prepared from the corresponding 2,3-diarylpyrroles by treatment with a strong base, such as sodium hydride, followed by alkylation using an alkyl halide, or other suitable alkylating reagent, such as methyl iodide,



R₆' = alkyl, benzyl or substituted benzyl

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Introduction of the α, α -bis (polyfluoro-alkyl) methanamine group is accomplished by reaction of the 2,3-diarylpyrrole with a fluorinated ketone imine, such as hexafluoroacetone imine. This reaction can be conducted in an inert solvent, such

as toluene, at temperatures from ambient to the reflux temperature of the solvent. Acidic catalysts such as AlCl₃, BF₃, p-toluenesulfonic acid, trifluoroacetic acid or the like are employed to increase the reaction rate. Reaction times are usually from less than one hour to 24 hours. The use of hexafluoroacetone imine in toluene at ambient temperature with catatylic amounts

of trifluoroacetic acid or aluminum chloride is preferred. When the polyfluorinated ketone imine employed is substituted on nitrogen by an R₇ group
(other than H), this reaction gives rise directly to
the corresponding substituted 4,5-diaryl-a,d-(polyfluoroalkyl)-lH-pyrrole-2-methanamines (R₇,= alkyl,

benzyl or substituted benzyl).

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Compounds in which one of R₄ and R₅ is H can be prepared by reduction of the oximes, which are in turn prepared by reaction of the corresponding 1-(4,5-diary1-1H-pyrrol-2-y1)polyfluoroalkanones with hydroxylamine.

$$\xrightarrow{\text{reduction}} \begin{array}{c} R_2 \\ R_3 \\ R_6 \end{array} \begin{array}{c} R_1 \\ CHR_4 \\ R_6 \end{array}$$

Thus, the 1-(4,5-diary1-1H-pyrrol-2-y1)-poly-fluoro-1-alkanones can be prepared from the corresponding 2,3-diary1pyrroles by treatment with a poly-fluorinated acid anhydride in the absence or presence of a base, such as N,N-dimethylaniline. The reaction can be run in any solvent which is inert to the reactants, at temperatures from -78° to the boiling point of the solvent, preferably at 0°C.

The preparation of the oximes is carried out by heating the polyfluoroalkanone in the presence of hydroxylamine hydrochloride and a base (such as an alkali metal acetate or alkoxide) in a polar solvent such as ethanol.

The reduction of the oxime is carried out by catalytic hydrogenation or by metal hydride reduction. Preferred conditions involve the use of lithium aluminum hydride in an ether solvent, such as diethyl ether or tetrahydrofuran at room temperature.

A similar procedure used in the preparation of 1-pheny1-2,2,2-trifluoroethylamine hydrochloride has been described in the literature [R. A. Shepard and S. E. Wentworth, J. Org. Chem., 32, 3197 (1967)].

$$\begin{array}{ccc}
 & & & \text{NOH} & & \text{NH}_2 \\
 & & & \text{C}_6 \text{H}_5 \text{CCF}_3 & \longrightarrow & \text{C}_6 \text{H}_5 \text{CHCF}_3 \cdot \text{HC1}
\end{array}$$

Compounds of Formula I of this invention with R_6 and/or $R_7 \neq H$ can alternatively be prepared by alkylation of the corresponding compounds with R_6 and/or $R_7 = H$. Alkylation can occur on either or both of the NH $_2$ or NH functionalities, depending on the conditions of the reaction. Often mixtures of alkylated products are obtained. These alkylations can be conducted in the presence or absence of a base, such as potassium carbonate,

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alkylating agent

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pyridine, triethylamine, potassium-t-butoxide, sodium hydride or the like. Examples of alkylating agents are methyl iodide and benzyl bromide.

Pharmaceutically suitable salts of the compounds of Formula I can be prepared by treatment of the 20 free base I with the appropriate acid.

In the following examples, all parts are by weight and temperatures are in degrees centigrade unless otherwise specified.

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Preparation 1

2,3-Diphenylpyrrole (Method A)

Dimethyl 4,5-diphenylpyrrole-2,3-dicarboxylate In a 2 i RB 3-neck flask with mechanical stirrer 5 and condenser was placed 76.7 g (0.31 mole) of desyl amine hydrochloride [Pschorr et al, Chem. Ber., 35, 2740 (1902)], 750 ml methanol, 88 g (0.62 mole) dimethyl acetylenedicarboxylate (freshly distilled) and 61 g (0.75 mole) anhydrous sodium acetate. The mixture 10 was heated at reflux for two hours. Then another 44 g (0.31 mole) of dimethyl acetylenedicarboxylate was added, and heating continued another two hours. While the reaction mixture was still at reflux, concentrated hydrochloric acid (~60 ml, to pH~2) was added dropwise. The mixture was heated at reflux another hour, then poured into 22 water containing 200 ml 10% sodium bicarbonate solution. With stirring, more socium bicarbonate was added until the solution was neutral. The gummy solic which precipitated was collected and washed with water. Trituration of this gummy material with ~500 ml of 50% aqueous ethanol gave a tan powdery solid, which was recrystallized from ~25% aqueous ethanol to give 65.5 g (63%) of white crystals, m.p. 25 191-2° [Lit. m.p. 185-7°; J. B. Hendrickson et al, J. Am. Chem. Soc., 86, 107 (1964)].

4,5-Diphenylpyrrole-2,3-dicarboxylic Acid To a mixture of 57.5 g (0.172 mole) of dimethyl 4,5-diphenylpyrrole-2,3-dicarboxylate in 350 ml methanol was added a solution of 71 g (1.78 mole) of 5 sodium hydroxide in 350 ml water. The mixture was heated at reflux for two hours, then cooled in an ice bath. The insoluble white crystals were collected and washed with cold methanol to give the bis sodium salt of the product. The still damp solid was dissolved in 10 12 cold water and acidified with conc. hydrochloric acid. The precipitated product was collected by filtration, washed with water containing ~1% hydrochloric acid, then air dried and finally dried in a vacuum oven at 100° to give 50.0 g (95%) of white 15 solid, m.p. 216-218° (dec., depends on heating rate).

C. 2,3-Diphenylpyrrole (Method A)

A mixture of 20 g (0.065 mole) of 4,5-diphenylpyrrole-2,3-dicarboxylic acid in 80 ml quinoline was
heated at reflux in an oil bath (bath~230°) until gas
evolution stopped (approx. one-half hour). The reaction
mixture was cooled and most of the quinoline was removed by distillation (bp 58° @ 0.2 mm). The partially
crystalline residue was chromatographed on 300 g Silic
AR CC-7, eluting with toluene to give 12 g (85%) of
faintly pink 2,3-diphenylpyrrole which could be further
purified by recrystallization from ethanol/water or by
sublimation (~125° @ 0.2 mm) to give white solid, m.p.
132-3°.

<u>Anal.</u> Calcd. for C₁₆H₁₃N: C, 87.64; H, 5.98; N, 6.39.

Found: C, 87.99; H, 5.86; U, 6.50.

Preparation 2

2,3-Diphenylpyrrole (Method B)

- A. Glyoxal mono(dimethylhydrazone) was prepared by the procedure of T. Severin and H. Poehlmann, Chem. Ber., 110, 491 (1977) to give 36.1 g (80%) of pale yellow liquid, bp 109° (22 mm); lit. bp 90° (16 mm).
 - B. 4-Dimethylhydrazono-1,2-diphenyl-2-buten-1-one

To a mixture of 19.6 g (0.1 mole) desoxybenzoin and 10 g (0.1 mole) of glyoxal mono(dimethylhydrazone) 10 in 100 ml ethanol was added dropwise a solution of sodium ethoxide prepared by dissolving 2.3 g (0.1 mole) sodium metal in 100 ml ethanol. The mixture was heated at reflux for one-half hour. TLC (90/10, toluene/ethyl 15 acetate) showed a small amount of starting desoxybenzoin, so 2.0 g (0.02 mole) of additional glyoxal mono (dimethylhydrazone) was added. Heating was continued another two hours. TLC at this time showed no starting material, and two clean close yellow product spots (isomers). The mixture was poured into 1 ℓ ice water then extracted with methylene chloride. The organic extracts were dried and concentrated on a rotary evaporator to give 28.7 g (100%) of yellow oil. The NMR showed the presence of two major N(CH₃)₂ containing materials (product isomers). 25 crude oil crystallized from isopropanol to give one pure isomer of product, 13.4 g (48%), pale yellow crystals, m.p. 131-2°.

Anal. Calcd. for C18H18N2O: C, 77.67; H, 6.52;

N, 10.06.

Found: C, 77.44; H, 6.46;

N, 10.17.

C. 2,3-Diphenylpyrrole (Method B)

A mixture of 3.1 g (0.011 mole) of 4-dimethylhydrazono-1,2-diphenyl-2-buten-1-one, 11.2 g (0.064 mole) sodium hydrosulfite in 75 ml ethanol and 37.5 ml 5 water was heated at reflux for three hours. The mixture was cooled and poured into 300 ml ice water. The white crystalline produce was collected, washed with water and air dried to give 1.9 g (79%), m.p. 130-1°, identical to product obtained via the decarboxylation, Method A.

Preparation 3

2,3 -Diphenyl-4-methylpyrrole

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A. Ethyl 4,5-diphenylpyrrole-3-carboxylate was prepared by a procedure similar to that used by A. M. van Leusen et al., Tet. Letters, 5337 (1972) for the preparation of the methyl ester. The ethyl ester was obtained as a white solid, m.p. 207-208.5° (methyl cyclohexane/toluene).

<u>Anal</u>. Calcd. for C₁₉H₁₇NO₂: C, 78.33; H, 5.88; 20 ____N, -4.81. ----

> Found: C, 77.92; H, 5.87;

25 B. 2,3-Diphenyl-4-methylpyrrole

To a stirred slurry of 0.76 g (20 mmoles) lithium aluminum hydride in 25 ml THF was added dropwise a solution of 0.58 g (2 mmoles) of ethyl 4,5-diphenylpyrrole-3-carboxylate in 5 ml THF. mixture was heated at reflux overnight. After cooling, 0.8 ml water, 2.4 ml 15% sodium hydroxide solution and 0.8 ml water were added dropwise. The solids were removed by filtration and the filtrate concentrated by rotary evaporation. The crystalline residue was purified by chromatography on 50 g silicic acid (CC-F),

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eluting with hexane/toluene (90/10) to give 0.25 g of product, m.p. 163-4°.

Anal. Calcd. for C₁₇H₁₅N: C, 87.51; H, 6.48; N, 6.00.

Found: C, 87.77; H, 6.60;

N, 5.89.

Preparation 4

2,3-Bis(4-fluorophenyl)-4-methyl-lH-pyrrole

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A. 3-(4-Fluorophenyl)-2-methyl-2-propen-1-al

To a solution of 124 g (1 mole) of 4-fluoro-benzaldehyde and 8 g (0.143 mole) of potassium hydroxide in 500 ml ethanol at room temperature was added dropwise a solution of 52.2 g (0.9 mole) of propionaldehyde in 100 ml ethanol. After stirring for 0.5 hour, the mixture was acidified with acetic acid and concentrated by rotary evaporation. The residue was partitioned between methylene chloride and water. The aqueous layer was extracted three times with additional methylene chloride. The combined organic layers were dried and concentrated. Distillation through a 12-inch vacuum jacketed column gave 113.5 g (77 %) of pale yellow low-melting crystalline product, b.p. 70-72°C (0.4-0.7 mm).

Anal. Calcd. for C₁₀H₉FO: C, 73.16; H, 5.53 Found: C, 72.89, 72.72; H, 5.66, 5.46.

B. 3-(4-Fluorophenyl)-2-methyl-2-propen-1-ol

To a solution of 113 g (0.69 mole) of 3-(4-fluoro-phenyl)-2-methyl-2-propen-1-al in 800 ml ethanol at 10°C was added in portions 13.1 g (0.345 mole) of sodium borohydride. After the addition was complete, the reaction mixture was stirred at room temperature overnight. The mixture was cooled in an ice bath while 350 ml of lN hydrochloric acid was added dropwise 35 to give a final pH of ~7. The mixture was diluted

with 500 ml water and extracted three times with methylene chloride. The organic extracts were dried and concentrated and the residue distilled to give 56.1 g (49%) of colorless liquid, b.p. 68-70°C (0.15 mm).

5 Anal. Calcd. for C₁₀H₁₁FO: C, 72.27; H, 6.67 Found: C, 72.30, 72.38;

H, 6.61, 6.62.

- C. <u>l-Chloro-3-(4-fluorophenyl)-2-methyl-2-propene</u>
 To a solution of 53.6 g (0.32 mole) of 3-(4-
- fluorophenyl)-2-methyl-2-propen-1-ol in 100 ml
 methylene chloride was added dropwise a solution of
 57.1 g (0.48 mole) of thionyl chloride in 100 ml
 methylene chloride. The reaction mixture was stirred
 at room temperature for 2 hours, then concentrated
- by rotary evaporation. The product was checked by NMR, then used crude in the reaction with ammonia.
 - D. 3-(4-Fluorophenyl)-2-methyl-2-propen-1-amine
 A quantity of 59.1 g (0.32 mole) of 1-chloro-3(4-fluorophenyl)-2-methyl-2-propene and 500 ml ethanol
- 20 was loaded in a pressure vessel. The vessel was cool-evacuated and 100 g of ammonia was added. The mixture was heated at 95° for 3 hours with shaking. The vessel was cooled, vented and the contents rinsed out with ethanol. The mixture was concentrated by
- 25 rotary evaporation. The residue was diluted with 1.5 l water and acidified with concentrated hydrochloric acid. This mixture was filtered to remove some insoluble solid (undissolved amine hydrochloride). The aqueous filtrate was extracted with ether to
- remove any non-basic impurities. The aqueous layer was combined with the insoluble solid and made basic with 5% sodium hydroxide solution. This was then extracted with ether and the ether extracts were dried and concentrated. Distillation of the residue gave
- 35 22.8 g (43%) of colorless liquid, b.p. 57°C (0.2 mm).

Anal. Calcd. for C₁₀H₁₂FN: C, 72.70; H, 7.32; N, 8.48

Found: C, 72.67, 72.59;

H, 7.48, 7.53

N, 8.31.

E. 4-Fluoro-N-[3-(4-fluorophenyl)-2-methyl-2-propenyl]benzamide

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To a vigorously stirred mixture of 19.8 g

- 10 (0.12 mole) of 3-(4-fluorophenyl)-2-methyl-2propen-1-amine and 30.2 g (0.36 mole) of sodium
 bicarbonate in 500 ml water at 5°C was added dropwise 22.2 g (0.14 mole) of 4-fluorobenzoyl chloride.
 The mixture was stirred another 3 hours at 5°C
- then at room temperature overnight. The white solid which had formed was collected, washed with saturated sodium bicarbonate solution, then with water, then with hexane, then air dried to give 33.4 g (97%) of product, m.p. 107-109°C.

20 <u>Anal.</u> Calcd. for C₁₇H₁₅F₂NO: C, 71.07; H, 5.26 N, 4.88

Found: C, 70.85; H, 5.48;

N, 4.70

Using the general procedure of N. Engel and
W. Steglich, Angew. Chem. Int. Ed. Engl., 17, 676
(1978), to a slurry at room temperature of 28.7 g
(0.1 mole) of 4-fluoro-N-[3-(4-fluorophenyl)-2-methyl2-propenyl]benzamide in 100 ml toluene containing 1 ml
DMF, stirred under nitrogen, with a dry ice condenser
attached, was added dropwise a solution of 39.6 g
(28.3 ml, 0.4 mole) of phosgene in 100 ml toluene.
The mixture was warmed slightly with a heat gun, then
stirred at room temperature overnight. The solution
was concentrated by rotary evaporation to give a
yellow oil. This was dissolved in 100 ml dry THF

(small amount of insoluble solid removed by decanting the solution) and the solution was added dropwise to a cool (15°) solution of 33.5 g (0.3 mole) of potassium t-butoxide in 150 ml DMSO. The dark purple 5 solution was stirred at ~20°C for 1 hour, then was poured into 1 liter ice water. This was extracted with ether and the ether layers backwashed with water. The ether layer was dried and concentrated and the residue was chromatographed on 10 900 g of silica gel, eluting with hexane containing 10-40% toluene, to give, after recrystallization from methyl cyclohexane, 10.8 g (40%) of white product, m.p. 126-7°C.

Anal. Calcd. for C₁₇H₁₃F₂N: C, 75.82; H, 4.87; Found: C, 75.87; H, 4.85; N. 5.13.

--Preparation 5 -

20 2,3-bis(4-Fluorophenyl)-1-methyl-1H-pyrrole

> To a mixture of 1.5 g (0.038 mole) of 60% sodium hydride dispersion and 100 ml DMSO was added dropwise a solution of 5.1 g (0.02 mole) of 2,3-bis(4-fluorophenyl)-lH-pyrrole in 25 ml

- DMSO. After the mixture was stirred one hour at room temperature, 5.6 g (0.04 mole) of methyl iodide was added dropwise. The mixture was stirred at room temperature overnight, then poured into water and extracted with ether.
- 30 The ether extracts were backwashed with water three times, then dried and concentrated. The crude solid was recrystallized from hexane to give 4.3 g of product, m.p. 129-129.5°.

Anal. Calcd. for C₁₇H₁₃F₂N: C, 75.82; H, 4.87;

N, 5.20

Found: C, 75.89, 75.78;

H, 4.98, 4.97;

N, 5.18, 5.10.

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Other 2,3-diarylpyrroles prepared by these procedures are given in Table I.

Table I 2,3-Diarylpyrroles

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R₂ R₁ R₁ R₃ R₆

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| | Prep. | R ₂ | <u>R</u> ₃ | R ₁ | R ₆ | m.p.°C | Yield (%) |
|----|-------|--|---|----------------|---|-------------|--------------|
| 15 | 6 | 4-C1C6H4 | 4-C1C ₆ H ₄ | Н | Н | 124-127 | 70 |
| Ť | · 7 | 4-FC ₆ H ₄ | 4-FC ₆ H ₄ | Н | H | 119.5-120.5 | 80 |
| | 8 | C6H5 | 3,4-diC1C ₆ H ₃ | * H | T. H = 17-70.7 | 112-113 | 28 |
| | 9 | 4-FC ₆ H ₄ | 4-BrC ₆ H ₄ | Н | -H | - 129-130 | 69 |
| | 10 | C6H5 | 3-pyridyl | Н | Н | 190-192 | 23 |
| 20 | 11 | 4-CH ₃ 0C ₆ H ₄ | 4-CH ₃ 0C ₆ H ₄ | Н | Н | oil | 64 |
| 20 | 12 | 4-CH3C6H4 | 4-CH ₃ C ₆ H ₄ | _ Н | Н | 128-129 | 83 |
| | 13 | 4-FC ₆ H ₄ | 4-(CH ₃)2NC ₆ H ₄ | Ή. | H | 200-201 | - 47 |
| | 14 | 4-FC ₆ H ₄ | 3-pyridyl | Ĥ | Н | 173-174 | 17 |
| | . 15 | 4-FC ₆ H ₄ | 4-CH_5C6H4 | Н | H | 164-165 | 71 . |
| 25 | 16 | 4-FC ₆ H ₄ | 4-CH_SO_C6H4 | H | H | 268-270 | 48 |
| 45 | 17 | 4-FC ₆ H ₄ | 4-FC ₆ H ₄ | H. | CH ₂ C ₆ H ₅ | 118-119 | 35 |

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Example 1

4,5-Bis (4-fluorophenyl)-α,α-bis (trifluoromethyl)lH-pyrrole-2-methanamine

A mixture of 3.85 g (0.015 mole) of 2,3-bis-(4-fluorophenyl)-lH-pyrrole and 3.0 g (0.018 'mole) of hexafluoroisopropylidenimine in 75 ml toluene was stirred at room temperature under a dry ice condenser for one hour, then heated at reflux for one hour. Since TLC indicated little 10 reaction, a quantity of 0.1 g of aluminum chloride was added to the cooled reaction mixture. The mixture was stirred at room temperature overnight. An additional quantity of 1.0 g (0.006 mole) of hexafluoroisopropylidenimine was added in 15 1 ml of cold toluene and, after two hours at -room temperature, another 0.1 g of aluminum chloride was added. The mixture was stirred at room temperature overnight, then concentrated by rotary evaporation. The residue was chromatographed on silica gel, eluting with toluene/ hexane mixtures, to give, after recrystallization from hexane, 4.6 g of product, m.p. 88-89°.

Anal. Calcd. for C₁₉H₁₂F₈N₂: C, 54.30; H, 2.88; N, 6.67.

Found: C, 54.25; H,2.86; N, 6.51.

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Example 2

4,5-Bis(4-fluorophenyl)-d-(trifluoromethyl)-lH-pyrrole-2-methanamine

A. 1-[4,5-Bis(4-fluorophenyl)-lH-pyrrol-2-yl]-2,2,2-trifluoroethanone

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To a solution of 2.5 g (0.012 mole) of trifluoro-acetic anhydride in 30 ml ether at 0° was added drop-wise a solution of 2.6 g (0.01 mole) of 2,3-bis(4-fluorophenyl)-lH-pyrrole and 1.5 g (0.012 mole) of N,N-dimethylaniline in 20 ml ether. The reaction mixture was stirred at 0° for 1.5 hours, then diluted with more ether, washed successively with water, 1N hydrochloric acid, then water again. The organic layer was dried and concentrated and the residue was purified by chromatography on silica gel, eluting with toluene, to give 2.7 g of white product, m.p. 211-212° (recrystallized from methyl cyclohexane/toluene).

Anal. Calcd. for $C_{18}H_{10}F_{5}N0$: C, 61.55; H, 2.87; N, 3.99

Found: C, 61.65; H, 3.13; N, 3.52.

B. 1-[4,5-Bis(4-fluorophenyl)-lH-pyrrol-2-yl]-2,2,2-trifluoroethanone, Oxime

A mixture of 0.42 g (6 mmoles) of hydroxylamine hydrochloride, 0.3 g (6 mmoles) of sodium methoxide and 1.05 g (3 mmoles) of the product from part A in 50 ml ethanol was heated at reflux overnight. An additional quantity of hydroxylamine, prepared from 0.42 g of hydroxylamine and 0.3 g of sodium methoxide, was added and heating was continued another three days. Another batch of hydroxylamine was added and heating continued overnight. This was repeated again, then the reaction mixture was poured into water and the white solid precipitate was collected and washed with water. Chromatography on silica gel, eluting with a

solvent containing 75% to 100% toluene and 25% to 0% hexane, gave 0.5 g of the oxime as a white solid, m.p. 215-216° (recrystallized from methyl cyclohexane/toluene).

5 <u>Anal</u>. Calcd. for C₁₈H₁₁F₅N₂O: C, 59.02; H, 3.03; N, 7.65. Found: C, 59.00; 59.07; H, 3.06, 3.13;

N, 7.65, 7.62.

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C. 4,5-Bis(4-fluorophenyl)-d-(trifluoromethyl)-lHpyrrole-2-methanamine

To a stirred mixture of 0.8 g (0.02 mole) of lithium aluminum hydride in 50 ml ether was added dropwise a solution of 3.7 g of the oxime from part B in 50 ml ether. The reaction mixture was stirred overnight at room temperature, then heated at reflux overnight. A quantity of 50 ml dry THF was added and heating at reflux was continued two more days. A solution of 8 g of sodium hydroxide in 15 ml water was added dropwise, then the mixture was diluted with more water and ether. The organic layer was separated after addition of lN hydrochloric acid to reduce the emulsion. The aqueous layer was extracted with more ether and with methylene chloride. The combined 25 organic layers were dried and concentrated. The crude product was purified by chromatography on silica gel, eluting with toluene:ethyl acetate (90:10) to give 1.1 g of pure product as an oil, which was characterized by TLC, IR, H- and F-NMR and mass spectrum. 30

> Mass Spectrum: Calcd. for C₁₈H₁₃F₅N₂: 352. Found: 352.

Other 4,5-diaryl-4,4-di(polyfluoroalkyl)-lH-pyrrole-2-methanamines that have been prepared by these procedures are given in Table II.

Table II

| 10 | Ex. | R ₁ | R ₂ | R ₃ | R ₄ | R ₅ | R ₆ | R ₇ | m.p.ºC | Yield (%) |
|-------|-----|----------------|---|--|-----------------|---------------------------|-----------------|-----------------|--------------|----------------|
| | | | — 4-FC ₆ Н ₄ | | CF ₃ | CF ₃ | CH ₃ | H | 129-130 | 75 |
| | 4 | | 4-CH30C6H4 | | CF3 | CF ₃ | Н | H | 120-120.5 | 37 |
| | 5 | | 4-FC ₆ H ₄ | | | CF3 | · | CH ₃ | 129-130 | 73 |
| | | | 4-FC ₆ H ₄ | 4-BrC6H4 | _ | CF ₃ | | н | 125-126 | 77 |
| 15 | _ | | 4-CH ₃ C ₆ H ₄ | 4-CH ₃ C ₅ H _A | | CF ₃ | • | Н | 143-144 | 71 |
| | | | | 4-FC ₆ H _A | _ | CF ₃ | | • | 154-155 | 83 |
| | | H | | 4-(CH ₃) ₂ NC ₆ H ₄ | | | | Η. | 155-156 | 87 |
| | 10 | | 4-C1C ₆ H ₄ | 4-C1C ₆ H ₄ | CF ₋ | CF ₃ | H | H | 119-119.5 | 82 |
| | 11 | Н | C ₆ H ₅ | C ₆ H ₅ | _ | .CF ₃ : | | | 124-125 | 90 |
| 20 | | | 6.5 4-FC ₆ H ₄ | 3-pyridyl | _ | CF ₃ | • | | 249-250 | 16 |
| | 13 | H | • | 4-FC ₆ H _A | | | | | 111-111.5 | —25 ··· |
| . • | | Н | 4-FC ₆ H ₄ | U T '. | | CF ₃ | | | 110-111 | 62 |
| | 14 | | 4-FC ₆ H ₄ | 4-FC ₆ H ₄ | _ | CF ₂ C1 | • = | _ | 94-95 | 5 |
| • - 0 | 15 | | | 4-FC ₆ H ₄ | | · · · ** · · · | | | 98-100* | |
| 25 | 16 | Н | 4-FC ₆ H ₄ | 4-FC ₆ H ₄ | ₩3 | 2'' | H. | | 70230 | |

*the compound of Example 16 was ~60-70% pure

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Following the procedure described, the following 4,5-diaryl-&-(polyfluoroalkyl)-lH-pyrrole-2-methan-amines can be prepared (Table III).

Table III

R₂ R₁ R₄ R₄ C-NHR₇

| | Ex. | <u>R</u> 1 | R ₂ · | R ₃ | R ₄ | R ₅ | R ₆ | R ₇ . |
|----|------|----------------------------------|--|----------------------------------|-----------------|---------------------------------|--|--|
| | 17 | Н | 4-FC6H4 | 4-FC ₆ H ₄ | | CF ₂ C1 | н | H |
| 15 | 18 | -CH ₂ CH ₃ | | | _ | CF ₃ | Н | Н |
| | -19 | н | 4-FC6H4 | | _ | _ | -CH ₂ CH ₂ CH ₃ | н - |
| | | | 4-FC6H4 | | CF ₃ | CF ₃ | Н | C6H13- |
| | | | 3-F, 4-C1C ₆ H ₃ | | | | | H |
| | | | | 4-FC6H4 | _ | CF ₃ | | CH ₃ |
| 20 | 23 | Н | | 4-CH3SO2C6H4 | _ | - | | Н Н |
| | -24 | | • • | 4-FC ₆ H ₄ | _ | _ | | н |
| | _25_ | Н | 4-FC6H4 | 4-FC6H4 | CF ₃ | CF ₃ | 4-NO2C6H4CH2- | H |
| | ~.26 | •н · | 4-FC ₆ H ₄ | 4-FC ₆ H ₄ | CF ₃ | CF ₃ | н 2042 | -CH ₂ C ₆ H ₅ |
| | 27 | | 4-FC ₆ H ₄ | | _ | CF_CF3 | | H |
| 25 | 28 | | | 4-FC6H4 | | CF ₂ CF ₃ | | H [*] |
| | 29 | | | 4-FC ₆ H ₄ | | CF ₃ | H | H |
| | 30 | H | 4-C2H50C6H4 | • - | | CF ₃ | H | H |
| | 31 | H | 4-(C2H5) 2NC6H4 | | _ | • | H | H . |
| | 32 | H | 4-CH ₃ SC ₆ H ₄ | 4-FC ₆ H ₄ | CF ₃ | • | CET 3 | H . |
| 30 | | ٠. | • | | • | | 4 1 4 | . : |

Dosage Forms

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The antiarthritic agents of this invention can be administered to treat arthritis by any means that produces contact of the active agent with the agent's site of action in the body of a mammal. They can be administered by any conventional means available for use in conjunction with pharmaceuticals; either as individual therapeutic agents or in a combintion of therapeutic agents. They can be administered alone, but are generally administered with a pharmaceutical carrier selected on the basis of the chosen route of administration and standard pharmaceutical practice.

The dosage administered will, of course, vary depending upon known factors such as the pharmacodynamic characteristics of the particular agent, and its mode and route of administration; age, health, and weight of the recipient; nature and extent of symptons, kind of concurrent treatment, frequency of treatment, and the effect desired. Usually a daily dosage of active ingredient can be about 0.01 to 40 milligrams per kilogram of body weight. Ordinarily 0.05 to 20, and preferably 0.1 to 4 milligrams per kilogram per day given in divided doses 2 to 4 times a day or in sustained release form is effective to obtain desired results.

Dosage forms (compositions) suitable for internal administration contain from about 1.0 milligram to about 500 milligrams of active ingredient per unit. In these pharmaceutical compositions the active ingredient will ordinarily be present in an amount of about 0.5 - 95% by weight based on the total weight of the composition.

The active ingredient can be administered orally in solid dosage forms, such as capsules, tablets, and powders, or in liquid dosage forms, such as elixirs, syrups, and suspensions; it can also be administered parenterally, in sterile liquid dosage forms.

Gelatin capsules contain the active ingredient and powdered carriers, such as lactose, sucrose, mannitol, starch, cellulose derivatives, magnesium stearate, stearic acid, and the like. Similar diluents can be used to make compressed tablets. Both tablets and capsules can be manufactured as sustained release products to provide for continuous release of medication over a period of hours. Compressed tablets can be sugar coated or film coated to mask any unpleasant taste and protect the tablet from the atmosphere, or enteric coating for selective disintegration in the gastro-intestinal tract.

Liquid dosage forms for oral administration can contain coloring and flavoring to increase patient acceptance.

In general, water, a suitable oil, saline,

aqueous dextrose (glucose), and related sugar solutions

and glycols such as propylene glycol or polyethylene
glycols are suitable carriers for parenteral solutions.

Solutions for parenteral administration contain preferably a water soluble salt of the active ingredient,
suitable stabilizing agents, and if necessary, buffer
substances. Antioxidizing agents such as sodium bisulfite, sodium sulfite, or ascorbic acid either alone
or combined are suitable stabilizing agents. Also used
are citric acid and its salts and sodium EDTA. In
addition parenteral solutions can contain preservatives,
such as benzalkonium chloride, methyl- or propylparaben and chlorobutanol.

Suitable pharmaceutical carriers are described in Remington's Pharmaceutical Sciences, E. W. Martin, a standard reference text in this field.

Useful pharmaceutical dosage forms for administration of the compounds of this invention can be illustrated as follows:

A large number of unit capsules are prepared by filling standard two-piece hard gelatin capsules each with 50 milligrams of powdered active ingredient 175 milligrams of lactose, 24 milligrams of talc, and 6 milligrams magnesium stearate.

A mixture of active ingredient in soybean oil is prepared and injected by means of a positive displacement pump into gelatin to form soft gelatin capsules containing 50 milligrams of the active ingredient. The capsules are washed in petroleum ether and dried.

Tablets

A large number of tablets are prepared by

conventional procedures so that the dosage unit is 50

milligrams of active ingredient, 0.2 milligrams of

colloidal silicon dioxide, 5 milligrams of magnesium

stearate, 275 milligrams of microcrystalline cel
lulose, 11 milligrams of cornstarch and 98.8 milli
grams of lactose. Appropriate coatings may be applied

to increase palatability or delay absorption.

-- Injectable

A parenteral composition suitable for

administration by injection is prepared by stirring

1.5% by weight of active ingredient in 10% by volume
propylene glycol and water. The solution is sterilized
by commonly used techniques.

. Suspension

administration so that each 5 milliliters contain
10 milligrams of finely divided active ingredient,
200 milligrams of sodium carboxymethyl cellulose,
5 milligrams of sodium benzoate, 1.0 grams of sorbitol
35 solution, U.S.P., and 0.025 milliliters of vanillin.

29 Injectable

A parenteral composition suitable for administration by injection is prepared by stirring 1% by weight of active ingredient in sodium chloride injection U.S.P. XV and sterilizing by commonly used techniques.

Use

activities of compounds in this series and standard drugs, a test was used based on a standard model of arthritis for which there is good correlation with human efficacy. The model is adjuvant-induced arthritis in rats. Federation Proceedings, Vol. 32, No. 2, 1973, "Models Used for the Study and Therapy of Rheumatoid Arthritis" - Symposium of the American Society for Pharmacology and Experimental Therapeutics - states "The rat polyarthritis produced by intradermal injection of a suspension of Mycobacterium tuberculosis is mineral oil (adjuvant) has been used extensively for the screening of drugs of potential use in rheumatoid arthritis."

compounds of this invention have shown activity in adjuvant-induced arthritis in rats which is widely recognized as a good model of human rheumatoid arthritis.

METHODS

Established Adjuvant-Induced Arthritis in Rats

Lewis (Wistar) male rats (Charles River Breeding

Laboratories, Wilmington, Mass.) weighing 175-220 grams

were injected subcutaneously with 0.1 ml of adjuvant in

the plantar area of the right hind paw. The adjuvant

was prepared by bead-milling, heat-killed, lyophilized

Mycobacterium butyricum (Difco \$0640) in light mineral

oil (Fisher Scientific Co. \$0-119 Paraffin Oil -

Saybolt Viscosity 125/135) 5 mg/ml. Twenty non-arthritic control rats were injected with mineral oil. The animals received water and Wayne Lab-Blox ad libitum*.

The rats were held for 14 days to allow the

development of polyarthritis. The volume of the

uninjected, left-hind paw of each rat was measured by

using a Ugo Basile Volume Differential Meter, Model 7101.

Adjuvant injected rats showing no evidence of arthritis

were discarded and the arthritic rats were distributed

into groups of 10 having equal mean paw volumes with

equal standard deviation. Non-arthritic (oil-injected)

control rats were distributed to 2 groups of 10.

Suspensions of test compounds were prepared for dosing

by bead-milling (4 mm glass beads in rubber stoppered

serum bottles) for 4-5 hours in aqueous 1% polyvinyl

alcohol, 5% gum acacia and 0.5% methylparaben.

Test compounds were given orally by gavage once daily for 7 days (days 14-20). The 2 groups of oil injected, non-arthritic control rats and the 2 groups of arthritic control rats received vehicle only for 7 days. Paw volumes (uninjected left hind paw) were measured 20 hours after the last dose (on day 21).

Percent decrease from control mean paw volume

25 was calculated with the following formula:

Arthritic Vehicle Control Arthritic Treatment

Mean Paw Volume (ml) Mean Paw Volume (ml) X 100=

Arthritic Vehicle Control Non-Arthritic Vehicle

Mean Paw Volume (ml) Control Mean Paw

Volume (ml)

& Decrease from Control Mean Paw Volume

^{*}while on a 10-hour light - 14 hour-dark cycle

Dose-response regression lines of the % decrease were plotted on semi-log paper and the $\rm ED_{50}$ % for decrease from control paw volume was estimated by inspection.

Phenylquinone Writhing Test

The phenylquinone writhing test, modified from Siegmund, et al., Proc. Soc. Exp. Biol. Med., 95, 729 (1957), was employed. A test compound suspended in 1% methylcellulose was given orally to fasted (17-21 hours) female white mice, 5-20 animals per double blind test. Aqueous (0.01% phenyl-p-benzoquinone) phenylquinone was injected intraperitoneally 24 minutes later using 0.20 ml per mouse. Commencing at 30 minutes after the oral administration of the test compound, the mice were observed for 10 minutes for a characteristic stretching or writhing syndrome which is indicative of pain induced by phenylquinone. The effective analgesic dose for 50% of the mice (ED₅₀) was calculated by the moving average method of Thompson, W. R., Bact. Rev., 11, 115-145 (1947).

Table IV

Antiarthritic and Analgesic Activity

5 10 4-FC6H4 н -4-FC6H4 4-FC6H4 CF₃ CF₃ 15 4-FC6H4 3 Н CF3 CF3 4-CH30C6H4 4-CH30C6H4 CF₃ CF₃ 4-FC₆H₄ 4-BrC6H4 CF₃ CF₃ H 4-CH3C6H4 4-CH-CHACAHA CF₃. CF₃. Н CF3. CF3 20 CH₃ -4-(CH₃)₂NC₆H₄-CF₃-CF₃-.4-C1C₆H₄ ... H 4-C1C6H4 CF₃ CF₃ 10 C6H5 CF₃ CF₃ 11 Н CF₃ CF₃ 3-pyridyl 12 4-FC6H4 CF₃ CF₃ 25 13 Н CF₃ CF₃ CH-14 4-FC₆H₄ H 4-FC6H4

CF₃ CF₂C1

30

15

Н

33
Table IV (continued)

| • | Ex. | Adjuvant Arthritic ED ₅₀ (mg/kg) | Analgesic Phenylquinone Writhing ED ₅₀ (mg/kg) |
|-------------------------|----------|---|---|
| 5. | 1 | 0.45 | 27 |
| | 2 | >25(2%) | - |
| | 3 | 2.6 | > 108 |
| | . 4 | 3 . 3 | 45 |
| _ | 5 | 0.8 | > 108 |
| 10 | 6 | 3.4 | > 108 |
| | 7 | > 9 (45%) | > 108 |
| | 8 | 0.77 | > 108 |
| | . 9 | 4.9 | > 108 |
| | 10 | 4.4 | >108 |
| 15 | 11 | 20 | >108 |
| | 12 | 12.5 | > 108 |
| managaristans a tilla e | | >25(24%) | >108 |
| - | 14 | 1.0 | > 108 |
| | 15 | 0.7 | . — whether it makes a second |
| 20 | | | ······································ |
| | ···(""in | dicates not tested.) | |
| | · 90 m · | | |
| 25 | | | |
| () · · · · | | | |
| • • | | · : | . • • |
| 30 | | | • |

1. A compound of the formula

$$\begin{array}{c|c}
R_2 \\
R_3 \\
N \\
R_4 \\
C-NHR_7
\end{array}$$
(1)

wherein

10

 $R_1 = H \text{ or } C_1 - C_2 \text{ alkyl};$

 R_2 and R_3 independently = 3-pyridyl

or

.

 $X = H, F, C1, Br, C_1-C_2$ alky1, C_1-C_2 alkoxy, $di(C_1-C_2)$ alky1) amino or $CH_3S(0)_n$ where n=0, 1 or 2; and

Y = H, F or CI;

with the proviso that when Y is F or Cl, then

X is F.or Cl;

 R_4 and R_5 independently = H, CF_3 , CF_2H , CF_2C1 , $CFCl_2$ or CF_2CF_3 , with the proviso that no more than one of R_4 and R_5 can be H; and the further proviso that no more than one of R_4 and R_5 can be

CF₂CF₃;

R₆ and R₇ independently = H, C₁-C₆
alkyl, benzyl or benzyl substituted with up
to two atoms selected from the group con---sisting of F, Cl, Br, NO₂, and CF₃; with
the proviso that when R₄ or R₅ is H, then
R₇ must be H also; or

a pharmaceutically suitable acid addition salt thereof.

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2. A compound of Claim I where $R_1 = H$ or

сн₃.

3. A compound of Claim 1 where

$$R_2$$
 and R_3 independently = X

where X = Br, C1, F, CH_3O or $(CH_3)_2N$ and Y = H.

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4. A compound of Claim 1 where R_4 and R_5 =

CF₃.

5. A compound of Claim 1 where $R_6 = H$ or

CH₃.

6. A compound of Claim 1 where $R_7 = H$ or

15 CH3.

7. A compound of Claim 1 where

$$R_1 = H \text{ or } CH_3;$$
 $R_2 \text{ and } R_3 \text{ independently } = X$

20

where $X = Br, -C1, F, -CH_30$ or $(CH_3)_2N$ and Y = H; and R_4 and $R_5 = CF_3$; $R_6 = H$ or CH_3 ; and $R_7 = H$ or CH_3 .

- 8. The compounds of claim 1, selected from 4,5-bis-(4-fluorophenyl)-d,d-di(trifluoromethyl)-1H-pyrrole-2-methanamine, 4,5-bis-(4-fluorophenyl)-N-methyl-d,d-di(trifluoromethyl)-1H-pyrrole-2-methanamine, and 4,5-bis-(4-fluorophenyl)-N,1-dimethyl-d,d'-di(trifluoromethyl)-1H-pyrrole-2-methanamine.
- 9. A pharmaceutical composition consisting essentially of a suitable pharmaceutical carrier and an effective antiinflammatory amount of a compound of claims 1 8.

10. A process for preparing a compound of Claim 1 which comprises

(a) contacting a 2,3-diarylpyrrole of the formula

 $R_3 \xrightarrow{R_2} R$

with a polyfluorinated ketone imine R₄C(R₅)=NR₇,

where R₁-R₇ are as previously defined (except
that R₄ and R₅ cannot equal H), in the absence
or presence of a suitable acidic catalyst; and
optionally:

- (b) when at least one of R₆ and R₇ are H, contacting 15 the product of step (a) with an alkylating agent; and optionally:
 - (c) converting the product of step (a) or (b) into a pharmaceutically suitable salt.
- 20 11. A process for preparing a compound of Claim 1 which comprises
 - (a) contacting a l-(4,5-diaryl-lH-pyrrol-2-yl)-polyfluoro-l-alkanone oxime of the formula

 $R_{3} = R_{4}$ $R_{4} \neq H$ $R_{6} = R_{4}$ $R_{6} = R_{4}$

with a reducing agent, and optionally;

- (b) contacting the product of step (a) with
- 30 an alkylating agent; and optionally,
 - (c) converting the product of step (a) or (b) into a pharmaceutically suitable salt.



EUROPEAN SEARCH REPORT

EP 81 10 2932

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